Chemical Reaction Modeling



ME 498 CF1 🛱 Fall 2016

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$$\frac{\partial}{\partial t}(\rho Y_i) + \nabla \cdot \left(\rho \vec{v} Y_i\right) = -\nabla \cdot \vec{J}_i + R_i + S_i$$

 Y_i = local mass fraction R_i = rate of production, chemical reaction S_i = rate of production, addition from dispersed phase J_i = diffusion flux

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$$\vec{J}_i = -\rho D_{i,m} \nabla Y_i - D_{T,i} \frac{\nabla T}{T}$$

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 Y_i = local mass fraction R_i = rate of production, chemical reaction S_i = rate of production, addition from dispersed phase J_i = diffusion flux

$$\vec{J}_i = -\left(\rho D_{i,m} + \frac{\mu_t}{\mathrm{Sc}_t}\right) \nabla Y_i - D_{T,i} \frac{\nabla T}{T}$$

$$\frac{\partial}{\partial t}(\rho Y_i) + \nabla \cdot \left(\rho \vec{v} Y_i\right) = -\nabla \cdot \vec{J}_i + R_i + S_i$$

$$R_{\text{kin},r} = \frac{A_r T^{\beta} \exp\left(\frac{E_r}{RT}\right)}{p_{r,d}^{N_{r,n}}} \prod_{n=1}^{n_{\text{max}}} p_n^{N_{r,n}}$$

Mixture Fractions Approach (Non-Premixed Combustion)

$$\frac{\partial}{\partial t} \left(\rho \bar{f} \right) + \nabla \cdot \left(\rho \vec{v} \bar{f} \right) = -\nabla \cdot \left(\frac{\mu_{\text{lam}} + \mu_{\text{turb}}}{\sigma_t} \nabla \bar{f} \right) + S_m$$

$$f = \frac{Z_i - Z_{i,\text{ox}}}{Z_{i,\text{fuel}} - Z_{i,\text{ox}}}$$

Reaction Progress Variable Approach (Premixed Combustion)

$$\frac{\partial}{\partial t}(\rho\bar{c}) + \nabla \cdot \left(\rho\bar{v}\bar{c}\right) = -\nabla \cdot \left(\frac{\mu_{\text{turb}}}{\text{Sc}_t}\nabla\bar{c}\right) + \rho S_c$$

$$c = \frac{\sum_{k} \alpha_{k} \left(Y_{k} - Y_{k}^{u}\right)}{\sum_{k} \alpha_{k} Y_{k}^{eq}} = \frac{Y_{c}}{Y_{c}^{eq}}$$

Composition PDF Transport Approach (Finite-Rate Chemistry)

$$\frac{\partial}{\partial t}(\rho P) + \frac{\partial}{\partial x_i}(\rho u_i P) + \frac{\partial}{\partial \psi_k}(\rho S_k P) = -\frac{\partial}{\partial x_i}\left[\rho\left\langle u_i''|\psi\right\rangle P\right] + \frac{\partial}{\partial \psi_k}\left[\rho\left\langle \frac{1}{\rho}\frac{\partial J_{i,k}}{\partial x_i}|\psi\right\rangle P\right]$$